Two Ways of Hydrogen Peroxide Formation in the Oxidative Inactivation of Cyanobacterial Glutamine Synthetase

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Z. Naturforsch. **52c**, 812–816 (1997); received August 12, 1997

Glutamine Synthetase, Protein Oxidation, Ammonia Effect, Photosynthetic Hydrogen Peroxide Production, Cyanobacterium, *Anabaena variabilis*

Using crude extracts from the cyanobacterium Anabaena variabilis glutamine synthetase (GS) activity was rapidly irreversibly reduced to about 60% during dark incubation ("spontaneous GS inactivation"). An additional decrease was observed by the addition of ammonia in the light ("ammonia-mediated inactivation"). Both effects were prevented by EDTA, MnCl₂ or catalase indicative of the involvement of H_2O_2 . This is a key intermediate in oxidative enzyme inactivation. In both spontaneous and ammonia-mediated GS inactivation H_2O_2 is produced in different ways. Spontaneous inactivation is prevented by depletion of reduced pyridine nucleotides which apparently donate electrons to produce H_2O_2 . Fractionation of the crude extract showed that the light-enhanced GS inactivation by ammonia required the presence of thylakoid membranes. The photosynthesis inhibitor DCMU decreased GS inactivation by ammonia. For the inactivation in the light apparently H_2O_2 is produced from superoxide during photosynthetic electron transport.

Introduction

Glutamine synthetase (GS) is a key enzyme in nitrogen assimilation and subject to different kinds of regulation. Long-term regulation is caused by repression of the glnA gene (encoding GS) if ammonia is present in the culture medium. This is the main type of GS regulation in cyanobacteria (Rowell et al., 1979; Flores and Herrero, 1994). Furthermore, in many prokaryotes the activity is down-regulated in a short-term effect following the addition of ammonia which is the result of a covalent GS modification by adenylylation (Stadtman, 1990b; Engelhardt and Klemme, 1982). Such a down-regulation of GS activity was also expected in cyanobacteria, but it was found not to exist. Only in Synechocystis PCC 6803 a shortterm GS inactivation was observed (Merida et al., 1991).

Recently, the influence of ammonia on GS activity in cyanobacteria was reinvestigated. In crude

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extracts incubated in the light a marked reduction of GS activity was found, and in *Synechocystis* PCC 6803 this effect was interpreted as ADP ribosylation (Silman *et al.*, 1995). On the other hand GS in crude extracts from *Anabaena variabilis* was shown to be oxidatively inactivated upon the addition of NADH (Martin *et al.*, 1997). The purified inactive GS did not exhibit any property indicative of ADP ribosylation or adenylylation. In this report we show that GS inactivation in crude extracts is caused by oxidation of GS and that hydrogen peroxide is involved in GS inactivation. Two different modes of H₂O₂ generation are discussed.

Materials and Methods

Anabaena variabilis ATCC 29413 was grown in Arnon medium without nitrate (Arnon *et al.*, 1974). Cultures were started by inoculating 200 ml of medium with a cell density equivalent to 1.5 μ g per ml of chlorophyll. The tubes were gassed with CO₂-enriched air (1.6% CO₂, v/v) and illuminated with 150 μ E m⁻² s⁻¹ fluorescent light at 30 °C.

Crude extracts were prepared from cells at the end of the log phase (1–2 days old cultures). Cells were collected by centrifugation (5 min at $6500 \times g$)

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and washed once in 50 mm Hepes, 4-(2-hydroxyethyl)-1-piperazine ethanesulfonic acid, pH 7.0. The sediment was resuspended in a threefold volume of the same buffer. Then the cells were passed twice through a French press at 65 MPa and centrifuged at $10~000 \times g$ for 5 min. The supernatant is the crude extract. It was incubated in a water bath at 25 °C with an illumination of $200~\mu E~m^{-2}~s^{-1}$. At the times indicated samples were assayed for GS activity.

GS was generally assayed by the transferase reaction according to (Orr *et al.*, 1981). The reaction mixture (340 µl) contained 40 mm Hepes, pH 7.0, 30 mm L-glutamine, 0.3 mm MnCl₂, 60 mm NH₂OH, 0.4 mm ADP and 20 mm Na₂HAsO₄ and an appropriate amount of GS. After a 10-min incubation at 37 °C the reaction was stopped by the addition of 660 µl stop mix (3.3% FeCl₃, 2.0% trichloroacetic acid and 0.25 N HCl). Absorption was measured at 540 nm. Biosynthetic activity was determined as described in (Martin *et al.*, 1997).

Chlorophyll was determined in methanol extracts using the extinction coefficient from (Mackinney, 1941). Protein was measured by the bicinchoninic acid (BCA) method (Smith *et al.*, 1985).

Data of typical experiments are shown unless the standard error is given. All experiments were repeated three or more times. The deviation from the mean did not exceed \pm 10%.

Results and Discussion

During incubation of crude extracts prepared with Hepes buffer GS activity was reduced to about 60% of the initial activity (Figs. 1, 2 and 3). This effect was called "spontaneous GS inactivation" and was completed within 30 to 60 min. GS inactivation was enhanced by the addition of ammonia (Fig. 1). At least 100 µм ammonia was necessary and saturation was reached at 4 mm; the reaction was completed within 30 min. Both effects are seen using either the assay based on transferase or biosynthetic activity. pH did not change during incubation. GS inactivation was not prevented by protease inhibitors like PMSF (phenylmethylsulfonyl fluoride), leupeptin or pepstatin. Only little GS degradation was detected in immuno blots of crude extracts using a polyclonal antiserum raised against GS indicating that proteolytic de-

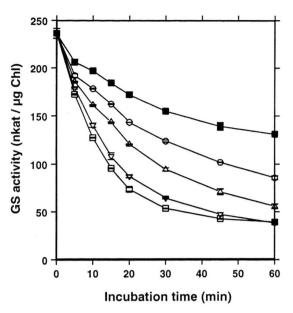


Fig. 1. Inactivation of GS by ammonia in crude extracts of *Anabaena variabilis*.

The experiment was started by adding ammonium chloride to 200 μ l of crude extract and incubating it at 25 °C with illumination at 200 μ E m⁻² s⁻¹. Incubations were stopped by starting the GS transferase assay. Standard deviations are derived from double assays. Incubations contained: \blacksquare , 0 mm ammonia; \bigcirc , 0.16 mm ammonia; \triangle , 0.63 mm ammonia; \bigcirc , 2.5 mm ammonia; \bigcirc , 10 mm ammonia.

gradation was not the reason for GS inactivation. The ammonia effect was absent when ammonia was substituted by glutamine or glutamate. This finding is not compatible with the view that GS is down-regulated by products of nitrogen assimilation as observed in *E. coli* or purple bacteria. Furthermore, no GS inactivation was observed with partially purified GS. Obviously, the crude extract contains substances mediating GS inactivation.

Both kinds of GS inactivation (spontaneous and ammonia-mediated) are impaired by EDTA, catalase and MnCl₂ (Table I). These substances are known to inhibit oxidative GS inactivation in *E. coli* (Levine *et al.*, 1981; Fucci *et al.*, 1983) and cyanobacteria (Martin *et al.*, 1997). Obviously, the observed GS inactivation in crude extracts is caused by oxidation. We have reported previously that in crude extracts GS oxidation requires NAD(P)H (Martin *et al.*, 1997), therefore a depletion of reduced pyridine dinucleotides should diminish GS

Table I. Effect of inhibitors on glutamine synthetase inactivation.

	GS activity (% of initial activity)		
Inhibitor	(-) NH ₄ Cl	(+) NH ₄ Cl, 4 mм	
EDTA, 1 mm	100	99	
MnCl ₂ , 400 μм	97	97	
Catalase, 8 mU / µl	101	98	
Control (water added)	72	51	

100 µl of crude extract were incubated in the absence or presence of ammonia for 30 min in the light at 25 °C and then assayed for GS activity. Catalase was from bovine liver (Boehringer Mannheim). Spontaneous GS inactivation is documented for the incubations without ammonia (1st col.), while the incubations containing ammonia show a combination of spontaneous and ammonia-enhanced GS inactivation (2nd col.).

Table II. Effect of depletion of reduced pyridine nucleotides on glutamine synthetase inactivation.

		GS activity (nkat/µg Chl)	
Additions	Incubation with:	30 min	60 min
 Alcohol dehydrogenase 	0 тм аттопіа	53	46
denydrogenase	4 mм ammonia Ratio	38 0.73	37 0.82
+ Alcohol dehydrogenase	0 тм аттопіа	90	82
, -8	4 mм ammonia Ratio	63 0.70	62 0.76

 $100\,\mu l$ of crude extracts were incubated in the absence or presence of ammonia for 30 min or 60 min in the light at 25 °C and then assayed for GS activity. Incubations with lowered content of reduced pyridine nucleotide contained 240 U / ml alcohol dehydrogenase (ADH) from yeast (Boehringer Mannheim) and 50 mM acetal-dehyde to form ethanol with consumption of NADH. Starting GS activity was 110 nkat / μg Chl. Spontaneous GS inactivation is given in the incubations without ammonia. The incubations containing ammonia show a combination of spontaneous and ammonia-enhanced GS inactivation. To get an estimate for the latter the ratio of the activity with added ammonia to that without ammonia is shown. Chl, chlorophyll.

inactivation. Table II shows that removal of reduced pyridine nucleotides by alcohol dehydrogenase only abolishes the spontaneous GS inactivation while the ammonia mediated GS inactivation is not affected.

While the spontaneous GS inactivation is independent of light the enhancement of GS inactivation by ammonia is only observed in the light (Fig. 2). To investigate the effect of light the influ-

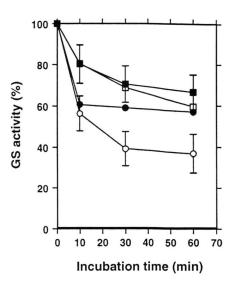


Fig. 2. Effect of light on glutamine synthetase inactivation.

Crude extracts were incubated in the light (200 µE m⁻² s⁻¹) or in darkness with or without the addition of ammonium chloride. Then GS activity was assayed. The values are normalized to 100% since they are derived from 23 (light) or 6 (dark) independent experiments. The initial GS activity was in the range of 180 to 240 nkat / µg chlorophyll. Incubations were: ■, 0 mM ammonia, dark; ●, 4 mM ammonia, dark; □, 0 mM ammonia, light; ○, 4 mM ammonia, light.

ence of photosynthesis was tested. After a short lag phase presence of the electron transport inhibitor DCMU [3-(3,4-dichlorophenyl)-1,1-dimethylurea] inhibited the ammonium-enhanced GS inactivation (Fig. 3) giving a result very similar to that obtained with ammonia in the dark (Fig. 2). The spontaneous GS inactivation was not influenced. To corroborate this finding thylakoid membranes were removed from the crude extract by ultracentrifugation which resulted in reduced GS inactivation by ammonia (Table III) while the spontaneous GS inactivation was even enhanced.

The enhancement of GS inactivation by ammonia is probably not caused by uncoupling of photophosphorylation which would result in increased NADP reduction and subsequent GS inactivation. In *Anabaena variabilis*, the half maximum concentration of uncoupling is 10 mm ammonia (Duane *et al.*, 1965) while GS inactivation by ammonia is saturated at 4 mm (Fig. 1). Moreover, the ammonia-mediated GS inactivation is not influenced by depletion of the NAD(P)H pool by alcohol dehydrogenase and acetaldehyde (Table II).

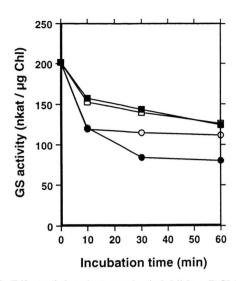


Fig. 3. Effect of the photosynthesis inhibitor DCMU on GS inactivation in crude extracts. Crude extracts were incubated in the light (200 μE m⁻² s⁻¹) or in darkness with or without the addition of ammonium chloride. Then GS activity was assayed. A typical experiment is shown. Incubations were: ■, 0 mM DCMU, 0 mM ammonia; ⊕, 0 mM DCMU, 4 mM ammonia; □ 100 μM DCMU, 0 mM ammonia; ○, 100 μM DCMU, 4 mM ammonia.

GS inactivation by ammonia is only observed when photosynthetic electron transport is operating, and it is assumed that H_2O_2 is produced during photosynthesis from superoxide (Takahashi

and Asada, 1988) generated mainly by reduction of O_2 through photosystem I. O_2 is much more soluble in the thylakoid membranes than in the cytosol. Accordingly, most of superoxide is generated in the membranes. To release this superoxide to the cytosol it must be protonated. Protonation is enhanced by ammonium penetrating the membrane which results in disproportionation of superoxide to O_2 and O_2 according to the following reaction (Takahashi and Asada, 1988):

 $2 O_2^- + 2 NH_4^+ \longrightarrow O_2 + H_2O_2 + 2 NH_3$ Oxidative GS inactivation was first detected in E. coli (Levine et al., 1981). In this organism inactivated GS was more rapidly degraded than the active enzyme. It was assumed that oxidative GS inactivation modifies the protein for subsequent proteolytic degradation (Stadtman 1990a). We described a similar system in Anabaena variabilis (Martin et al., 1997); recently ubiquitin was detected in Anabaena which is also involved in protein degradation (Durner and Böger, 1995). Irreversible oxidative inactivation was reported for cyanobacterial nitrogenase (Bagchi et al., 1991, Durner et al., 1996) and protein degradation was detected on SDS gels. These examples indicate that protein oxidation is an important step to start cyanobacterial protein degradation.

Table III. Inactivation of glutamine synthetase in fractions prepared by ultracentrifugation.

	Treatment	Incubation with:	GS activity (% of initial activity)		Activity at start
No.			30 min	60 min	(nkat / μg protein)
1 2 3	Supernatant	0 mм ammonia 4 mм ammonia Ratio	54 44 0.83	33 28 0.84	3.3 3.3
4 5 6	Pellet, containing the membranes	0 mм ammonia 4 mм ammonia Ratio	79 58 0.74	68 39 0.58	2.1 2.1
7 8 9	Control without ultracentrifugation	0 mм ammonia 4 mм ammonia Ratio	68 47 0.69	56 39 0.70	4.0 4.0

7.5 ml of crude extract were centrifuged at $170,000 \times g$ for 1 h at 4 °C. The sediment was resuspended in 1.5 ml of 50 mm Hepes, pH 7.0. $100 \,\mu$ l of the extract were incubated in the light at 25 °C in the absence or presence of ammonia for 30 min or 60 min, respectively. Then a 5- μ l sample was used to assay GS activity. Spontaneous GS inactivation is seen by the lines 1, 4, 7 (incubations without ammonia). The incubations containing ammonia (lines 2, 5, 8) show a combination of spontaneous and ammonia-enhanced GS inactivation. To estimate the latter inactivation the activity ratio with added ammonia to that without ammonia is shown (lines 3, 6, 9).

Acknowledgements

This study was supported by the Deutsche Forschungsgemeinschaft through its Sonder-

forschungsbereich 248 "Stoffhaushalt des Bodensees" and by the Fonds der Chemischen Industrie.

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